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Three States of

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Matter \ " The

*Creation of
Chemistry - The
Fundamental*

Laws: Crash

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~~Measurement~~

States of matter

| States of

matter and

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*intermolecular
forces |
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the Particle*

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nature of matter

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Changes of

States of Matter

Chemical and

Physical Changes

States of Matter

: Solid Liquid

Gas 10 Amazing

Experiments with

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Water And

Pure Substances
and Mixtures!

(Classification
of Matter)

Physical and
Chemical Changes

Understanding
Atoms, elements,
and molecules

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What is matter?

— HD Properties
Change Chapter
of Matter

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Physical and

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Changes States

of Matter—

Solids, Liquids,

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Elements \u0026

Compounds,

Classification

of Matter,

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Two Chemicals

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atmosphere,
Change Chapter
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Matter Chemistry
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1.2

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Chapter 1

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Section 2 MATTER

and Its

Change Chapter
Properties Matter

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- Anything that

Answers
has MASS and

Stoichiometry
VOLUME. ATOMS

and MOLECULES

the basic

building blocks

of matter.

ATOM-- the

smallest unit of

an element that

still maintains

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Change Chapter
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the properties
that element.
ELEMENT- a pure
substance made
up of only one
type of atom.

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Change Chapter 6
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Section 6.3

Periodic Trends

In your

textbook, read

about atomic

radius and ionic

radius. Circle

the letter of

the choice that

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best completes
the statement or
answers the
question. 1.

Atomic radii
cannot be
measured ...

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ter-Properties

and Changes

Solutions Manual

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3 35 Section 3.1

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Answers Problem-

Solving Lab 1.

Stoichiometry

Explain why the

flow of a

compressed gas

must be

controlled for

practical and

safe use.

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Table and

Periodic Law

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Development of
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1. octaves 2.
eight 3. nine 4.
accepted 5.
Dmitri Mendeleev
6. atomic mass
7. elements 8.
atomic number 9.
Henry Moseley
10. protons 11.
periodic law 12.
properties 13.
14 ...

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5 25 Electrons

in Atoms

Reviewing

Vocabulary Match

the definition

in Column A with

the term in

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Matter B. And Column
A Column B 1.
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The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

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answer key. some
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energy changes
(that is, energy
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absor.. A system
that can
exchange neither
energy nor
matter with its...
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extensive

property.

intensive

property.

extensive

properties

(examples)

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intensive
property
(examples) a
property that
depends on the
amount of matter
in a sample. a
property that
depends on the
type of matter
in a sample
not... mass,
volume, length,
weight.

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and change

chapter 2

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11 31 number of

moles of known

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substance C H C
H C H O O O CO
CO CO H O H O H
O number of
moles of unknown
substance O CO H
O C H CO H O C H
O H O C H O CO
use mole ratio 5
mol O 2 1 mol C
3 H 8 3 mol CO 2
1 mol C 3 H 8 4
mol H 2 O 1 mol
C 3 H 8 1 mol C

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3 H 8 5 mol 0 2

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MATH SKILLS

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MASTER 16

A “chemical change” is a change that produces matter with a different composition than the original matter. Chemical Change A change

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in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of
Compounds Quite
different
properties than
their component

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elements.
Change Chapter
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